

$\mu\text{g/ml}$. The medium was poured into petri dishes and left to harden overnight at room temp. The surface of the agar was then inoculated with the test organisms (0.02 ml of standard suspension). The inoculated plates together with the appropriate organism controls were incubated for 3 days at 37° in the case of *P. ovale* and up to 5 days in the case of *P. orbiculare*. MIC's were detd by observing the lowest concn which inhibited growth under the prescribed condns.

Suspension Technique. Each test compd (0.1 g) was dissolved or suspended in Tween 40 (2 ml) and the vol made up to 100 ml with sterile dist H₂O. A sample was inoculated with *P. ovale* (0.1 ml of standard suspension contg 10⁶ organisms/ml) and stored at room temp for 1 hr. The test samples and appropriate controls were plated out on petri dishes of Dixon's medium² and incubated at 37°

for 3 days. Activity of the compds was assessed on the growth observed.

References

- (1) J. Lodder, "The Yeasts; A taxonomic study," North Holland Publishing Co., Amsterdam, 1970, pp. 1166-1186.
- (2) N. J. Van Abbé, *J. Soc. Cosmet. Chem.*, **15**, 609 (1964).
- (3) T. H. Sternberg and F. M. Keddie, *Arch. Dermatol.*, **84**, 999 (1961).
- (4) R. C. Burke, *J. Invest. Dermatol.*, **36**, 389 (1961).
- (5) J. E. Hogan, Ph.D. Thesis, University of London, 1970.
- (6) J. K. Sugden, Ph.D. Thesis, University of London, 1964.
- (7) C. Caldo, *Chim. Ind. (Milan)*, **44**, 753 (1962).

New Compounds

A Rapid, Convenient Preparative Procedure for Phenethylamines

Edgar F. Kiefer

Department of Chemistry, University of Hawaii, Honolulu, Hawaii 96822. Received August 6, 1971

In view of the very broad pharmacological utility of substituted 2-phenylethylamines, we wish to contribute a synthetic procedure which, because of its versatility and convenience, may find considerable use. Although based entirely on standard synthetic methods, the overall scheme is specifically tailored to the properties of the benzylic intermediates involved, and eliminates the need for isolation of intermediates and other time-consuming operations. The procedure is described for the *p*-methoxy derivative; it is also applicable without substantive modification to other ring alkoxy-, alkyl-, and halogen-substituted phenethylamines.

Experimental Section

4-Methoxyphenylethylamine Hydrochloride. *p*-Anisyl alcohol (100 g, 0.725 mole) was shaken with 500 ml of concd HCl for 2 min. The org phase was washed with H₂O, 5% NaHCO₃, and H₂O, then added over 40 min to a stirred slurry of 49 g (1.0 mole) of NaCN in 400 ml of DMSO,¹ with ice-water cooling to maintain the temp at 35-40°. After addn was complete, the cooling bath was removed, the mixt was stirred for 90 min and then added to 300 ml of H₂O, and the small upper phase sepd. The aq DMSO layer was extd with two 100-ml portions of Et₂O, which were combined with the product layer, and the whole was washed once with H₂O and dried (MgSO₄).

A dry flask was charged with ca. 600 ml of abs Et₂O and chilled in ice as 80 g (0.6 mole) of anhyd AlCl₃ was added portionwise, followed by 23 g (0.6 mole) of LAH.^{2†} The dried Et₂O soln of crude *p*-methoxyphenylacetone nitrile was added at such a rate as to maintain gentle reflux without external heat (ca. 1 hr). The mixt was stirred for 2 hr, then chilled in ice, and treated dropwise with 25 ml of H₂O followed by 250 ml of 20% of aq NaOH, with periodic addn of Et₂O through the condenser to replenish losses and facilitate stirring. The resulting voluminous, granular ppt of NaCl and LiCl and aluminate was removed by filtration, washed well with Et₂O, and discarded. The filtrate was mixed with one-third its vol of abs EtOH and 60 ml of concd HCl was added slowly with continuous swirling and ice cooling. After chilling to 0°, the cryst amine hydrochloride was collected, 101 g, mp 212-214°, identified by mass spectroscopy [*m/e* 122, 30, 121, 28, 151 (M⁺)]. The overall yield was 75%

†LAH alone and other metal hydride reagents are unsatisfactory for the reduction of benzylic nitriles to amines.

from anisyl alcohol. The hydrochloride may be recrystd from Et₂O-EtOH or *i*-PrOH.

***N*-Methyl-*p*-methoxyphenylethylamine Hydrochloride.** *p*-Methoxyphenylethylamine, generated from 100 g (0.536 mole) of the hydrochloride by stirring with concd aq NaOH, was treated with 100 ml of PhH and 70 g (0.66 mole) of PhCHO. A mildly exothermic reaction began at once. The mixt was heated under reflux until no more H₂O was present in the condensate (ca. 1 hr), then, without cooling, an attached Dean-Stark trap was removed and a soln of 82 g (0.65 mole) of Me₂SO₄³ in 200 ml of PhH was added through the condenser at such a rate as to maintain reflux (15 min). The 2-phase mixt was heated for 90 min on the steam bath, cooled slightly, treated with 200 ml of H₂O, and heated for an addl 20 min. After cooling in ice, the aq layer was washed twice with Et₂O to remove unreacted PhCHO and made strongly basic with 50% aq NaOH. Two Et₂O exts of the basic aq phase were added to the amine layer which sepd, and the resulting soln was evacd at the aspirator for 30 min, leaving 90 g (102%) of crude *N*-methyl-*p*-methoxyphenylethylamine. This material was dissolved in 500 ml of 20% abs EtOH-Et₂O and treated with 50 ml of concd HCl with swirling and cooling to yield the white, cryst hydrochloride, which was washed thoroughly with ice-cold 20% EtOH-Et₂O and dried, mp 185.5-186.5°, identified by mass spectroscopy [*m/e* 121, 44, 165 (M⁺)]. The yield was 83 g (77%).

References

- (1) R. A. Smiley and C. Arnold, *J. Org. Chem.*, **25**, 257 (1960).
- (2) R. F. Nystrom, *J. Amer. Chem. Soc.*, **77**, 2544 (1955).
- (3) J. J. Lucier, A. D. Harris, and P. S. Korosec, *Org. Syn.*, **44**, 72 (1964).

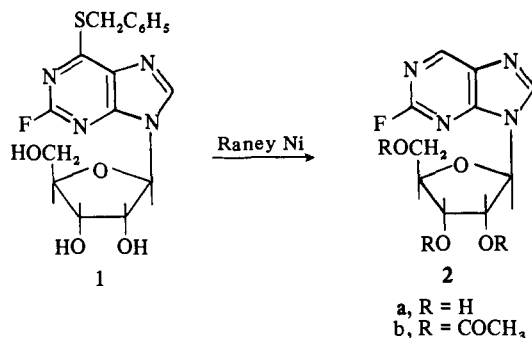
Synthesis of 2-Fluoro-9- β -D-ribofuranosylpurine (2-Fluoronebularine)

Masajiro Kawana, Robert J. Rousseau,* and Roland K. Robins

ICN Nucleic Acid Research Institute, Irvine, California 92664. Received August 16, 1971

The antibiotic nebularine (9- β -D-ribofuranosylpurine) has shown tuberculostatic,¹ antimitotic,² and anticancer activity.^{2,3} The mode of action has been proposed to be in the purine biosynthetic pathway.^{4,5} It has limited usefulness because of its high toxicity.^{2,6,7}

We wish to report the synthesis of 2-fluoronebularine (2a). Synthesis of the title compound 2a was accomplished by removal of the benzylthio group from 6-benzylthio-2-fluoronebularine (1)⁸ with Raney Ni.



Experimental Section†

2-Fluoro-9-(β -D-ribofuranosyl)purine (2a). To a boiling soln of 1.64 g (4 mmoles) of $1 \cdot \text{H}_2\text{O}$ in 30 ml of EtOH was added a suspension of 14 g of Raney Ni in 60 ml of EtOH. The mixt was refluxed for 40 min with stirring. The catalyst was removed by filtration with a Celite pad and washed thoroughly with boiling EtOH. The combined filtrate and washings were concd to ca. 10 ml *in vacuo*. The undissolved material was removed by filtration with charcoal and the filtrate was evapd to dryness *in vacuo* to give 590 mg (55%) of crude **2a** as a foam. The product was chromatogd on a silica gel column with EtOAc-EtOH (95:5, v/v). Evapn of the solvents gave a colorless foam, which was crystd from EtOAc contg a small amt of MeOH: mp 144.5–146°; $[\alpha]^{25}_D -32.3^\circ$ (*c* 1, H_2O); uv $\lambda_{\text{max}}^{\text{pH 11}}$ 263 nm (ϵ 8000), $\lambda_{\text{max}}^{\text{pH 11}}$ 264 (7500), $\lambda_{\text{max}}^{\text{MeOH}}$ 264 (8000); nmr (DMSO- d_6 - D_2O) δ 6.00 (d, $J_{1,2} = 5.4$ Hz, 1 H, H_1), 8.83 (s, 1 H, H_8), 9.06 (d, $J_{\text{H}_8\text{F}} = 1.2$ Hz, 1 H, H_8); nmr for F (DMSO- d_6) –24.8 ppm. *Anal.* Calcd for $\text{C}_{10}\text{H}_{11}\text{FN}_4\text{O}_4$: C, H, N.

9-(2,3,5-Tri-*O*-acetyl- β -D-ribofuranosyl)-2-fluoropurine (2b). To a stirred soln of 50 ml of 48–50% HBF_4 was added 3.9 g (0.01 mole) of 9-(2,3,5-tri-*O*-acetyl- β -D-ribofuranosyl)-2-aminopurine⁹ at –20 to –25°. To this mixt was added a soln of 2.1 g (0.03 mole) of NaNO_2 in 4 ml of H_2O over a period of 10 min. The reaction mixt was stirred at the same temp for another 15 min and 50 ml of EtOH (precooled below –20°) was then added. The mixt was neutralized with ca. 26 ml of concd NH_4OH to pH 6 below –15°. The resulting ppt was removed by filtration and washed with 50 ml of cold EtOH. The combined filtrate and washings were concd to ca. 50 ml at 30–35° *in vacuo*, and the soln was extd with two 150-ml portions of CH_2Cl_2 . The exts were washed with, successively, 50 ml of H_2O , 50 ml of 1% NaHCO_3 , and two 50-ml portions of H_2O , and then dried (MgSO_4). Evapn of the solvents gave 2.9 g (73%) of crude **2b** as a gummy material. This product was chromatogd on a silica gel column (130 g, 4 × 30 cm) using EtOAc-heptane (7:3, v/v), and 250-ml fractions were collected. Fractions 26–35 contd 960 mg (24%) of **2b**, which was contaminated with a trace of impurity. Fractions 36–67 were combined and evapn of the solvents gave 1.07 g (27%) of analytically pure **2b** as a glass: $[\alpha]^{25}_D -3.5^\circ$ (*c* 2.88, CHCl_3); uv $\lambda_{\text{max}}^{\text{pH 11}}$ 264 nm (ϵ 7300), $\lambda_{\text{max}}^{\text{pH 11}}$ 265 (7200), nmr (DMSO- d_6) δ 6.33 (d, $J_{1,2} = 4.8$ Hz, 1 H, H_1), 8.83 (s, 1 H, H_8), 9.13 (d, $J_{\text{H}_8\text{F}} = 1.2$ Hz, 1 H, H_8); nmr for F (DMSO- d_6) –27.7 ppm. *Anal.* Calcd for $\text{C}_{16}\text{H}_{17}\text{FN}_4\text{O}_7$: C, H, N.

Attempts to remove the Ac blocking groups of **2b** with EtOH- NH_3 at 4° occurred with concomitant displacement of the F at and formation of 9- β -D-ribofuranosyl-2-aminopurine.

References

- (1) N. Löfgren and B. Luning, *Acta Chem. Scand.*, **7**, 225 (1953).
- (2) J. J. Biesele, M. C. Slautterback, and M. Margolis, *Cancer (Philadelphia)*, **8**, 87 (1955).

†Melting points were taken on a Thomas-Hoover capillary melting point apparatus and are uncorr. Microanalyses were performed by Heterocyclic Chemical Corp., Harrisonville, Mo. Where analyses are indicated only by symbols of the elements, analytical results obtained for those elements were within $\pm 0.4\%$ of the theoretical values. The F nmr spectra were run with 1% CF_3COOH as an external standard.

- (3) G. B. Brown and V. S. Weliky, *J. Biol. Chem.*, **204**, 1019 (1953).
- (4) M. P. Gordon, D. I. Magrath, and G. B. Brown, *J. Amer. Chem. Soc.*, **79**, 3256 (1957).
- (5) R. J. Winzler, W. Wells, J. Shapiro, A. D. Williams, I. Bornstein, M. J. Burr, and W. R. Best, *Cancer Res.*, **19**, 377 (1959).
- (6) A. P. Truant and H. E. D'Amato, *Fed. Proc., Fed. Amer. Soc. Exp. Biol.*, **14**, 391 (1955).
- (7) M. P. Gordon and G. B. Brown, *J. Biol. Chem.*, **220**, 927 (1956).
- (8) J. F. Gerster and R. K. Robins, *J. Amer. Chem. Soc.*, **87**, 3752 (1965).
- (9) W. W. Zorbach and R. S. Tipson, "Synthetic Procedures in Nucleic Acid Chemistry," Interscience Publishers, New York, N. Y., 1968, p 244.

4-[(Aminoxy)methyl]thiazole Dihydrochloride

Glenn H. Hamor

Department of Biomedical Chemistry, School of Pharmacy,
University of Southern California, Los Angeles, California 90007.
Received August 27, 1971

Recent reports showing the title compound (I) to possess potent *in vitro* and *in vivo* inhibition of specific histidine decarboxylase¹ and to markedly lower rat brain histamine,^{2,3} prompt me to report its synthesis. By literature methods,⁴⁻⁶ alkylation of *N*-hydroxyphthalimide with 4-chloromethylthiazole, followed by hydrazinolysis, gave I.

Experimental Section†

***N*-(4-Thiazolylmethoxy)phthalimide (II).** To a soln of 34.9 g (0.214 mole) of *N*-hydroxyphthalimide in 250 ml of MeCN and 43.3 g (0.428 mole) of Et_3N was added 36.3 g (0.214 mole) of 4-chloromethylthiazole hydrochloride⁷ and refluxed for 6 hr. After cooling, the cryst ppt was filtered, washed with a little MeCN and thoroughly with H_2O , and dried, to yield 24 g (43%) of cryst product, mp 158–159° (EtOH). *Anal.* ($\text{C}_{12}\text{H}_8\text{N}_2\text{O}_3\text{S}$) C, H, N.

4-[(Aminoxy)methyl]thiazole Dihydrochloride (I). II (13 g, 0.05 mole) was refluxed for 2 hr with 2.5 g (0.05 mole) of hydrazine hydrate, 99–100%, in 150 ml of anhyd EtOH. After cooling and removal by filtration of pptd phthalhydrazide, ethanolic HCl was added to the filtrate, and the cryst solid was filtered off and dried. Recrystn from EtOH-Et₂O gave 8.5 g (83%) of white solid, mp 176–177° dec. *Anal.* ($\text{C}_4\text{H}_6\text{N}_2\text{OS} \cdot 2\text{HCl}$) C, H, N.

References

- (1) D. Aures, G. H. Hamor, W. G. Clark, and S. S. Laws, 4th International Pharmacology Congress, Basel, 1969, p 179.
- (2) M. K. Menon, D. Aures, and W. G. Clark, *Pharmacologist*, **12**, 205 (1970).
- (3) K. M. Taylor and S. H. Snyder, *Science*, **172**, 1037 (1971).
- (4) A. F. McKay, D. L. Garmaise, G. Y. Paris, and S. Gelblum, *Can. J. Chem.*, **38**, 343 (1960).
- (5) D. G. Martin, E. L. Schuman, W. Veldkamp, and H. Keasling, *J. Med. Chem.*, **8**, 455 (1965).
- (6) D. J. Drain, J. G. Howes, and H. W. R. Williams, British Patent 984,305 (1965); *Chem. Abstr.*, **62**, 14572 (1965).
- (7) W. T. Caldwell and S. M. Fox, *J. Amer. Chem. Soc.*, **73**, 2935 (1951).

†Melting points were detd with a Fisher-Johns app and are uncorr. Ir spectra (Nujol mull) were measured on a Perkin-Elmer infracord 137 spectrometer. Absorption bands were as expected. Elemental anal. were performed by Elek Microanalytical Laboratories, Torrance, Calif. Where analyses are indicated only by symbols of the elements, analytical results obtained for those elements were within $\pm 0.4\%$ of the theoretical values.